

Lista publikacji

- październik 2011

I. Prace oryginalne (rozdział w książkach zbiorowych, artykuły w czasopismach):

1. Norek M., Stępniewski W. J., **Polański M.**, Zasada D., Bojar Z., Bystrzycki J., **A comparative study on the hydrogen absorption of thin films at room temperature deposited on non-porous glass substrate and nano-porous anodic aluminum oxide (AAO) template**, International Journal of Hydrogen Energy 36: 11777-11784.
2. Norek M., Nielsen T. K., **Polański M.**, Kunce I., Płocioski T., Jaroszewicz L. R., Cerenius Y., Jensen T. R., Bystrzycki J., **Synthesis and decomposition mechanisms of ternary Mg₂CoH₅ studied using in-situ synchrotron X-ray diffraction**, International Journal of Hydrogen Energy 36: 10760-10770.
3. Nielsen T. K., **Polański M.**, Zasada D., Javadian P., Besenbacher F., Bystrzycki J., Skibsted J., Jensen T. R., **Improved hydrogen storage kinetics of nanoconfined NaAlH₄ catalyzed with TiCl₃ nanoparticles**, ACS Nano 5: 4056-4064.
4. **Polański M.**, Bystrzycki J., Varin R. A., Płociński T., Pisarek M., 2011, **The effect of chromium (III) oxide (Cr₂O₃) nanopowder on the microstructure and cyclic hydrogen storage behavior of magnesium hydride (MgH₂)**, Journal of Alloys and Compounds 509: 2386-2391.
5. Varin R.A., Zbroniec L., **Polański M.**, Bystrzycki J., 2011, **A review of recent advances on the effects of microstructural refinement and nano-catalytic additives on the hydrogen storage properties of metal and complex Hydrides**, Energies 4: 1-25.
6. **Polański M.**, Bystrzycki J., Varin R. A., Płociński T., 2011, **Rapid hydrogenation at 30°C of magnesium (Mg) and iron (Fe) nanocomposite obtained through a decomposition of Mg₂FeH₆ precursor**, International Journal of Hydrogen Energy 36: 1059-1065.
7. Varin R. A., Zaranski Z., Czujko T., **Polański M.**, Wronski Z. S., 2011, **The composites of magnesium hydride and iron-titanium intermetallic**, International Journal of Hydrogen Energy 36: 1177-1183.
8. **Polanski M.**, Nielsen T. K., Cerenius Y., Bystrzycki J., Jensen T. R., 2010, **Synthesis and decomposition mechanisms of Mg₂FeH₆ studied by in-situ synchrotron X-ray diffraction and high-pressure DSC**, International Journal of Hydrogen Energy 35: 3578-3582.
9. Varin R. A., Jang M., Polanski M., 2010, **The effects of ball milling and molar ratio of LiH on the hydrogen storage properties of nanocrystalline lithium amide and lithium hydride (LiNH₂+LiH) system**, Journal of Alloys and Compounds 491: 658-667.
10. **Polanski M.**, Płociński T., Kunce I., Bystrzycki J., 2010, **Dynamic synthesis of ternary Mg₂FeH₆**, International Journal of Hydrogen Energy 35: 1257-1266.
11. **Polański M.**, Bystrzycki J., 2009, **Comparative studies of the influence of different nano-sized metal oxides on the hydrogen sorption properties of magnesium hydride**, Journal of Alloys and Compounds 486(1-2): 697-701.

12. Polański M., Bystrzycki J., 2009, *The influence of different additives on the solid-state reaction of magnesium hydride (MgH₂) with Si*, International Journal of Hydrogen Energy 34(18): 7692-7699.
13. Bystrzycki J., Polański M., Plocinski T., 2009, *Nano-engineering approach to destabilization of magnesium hydride (MgH₂) by solid-state reaction with Si*, Journal of Nanoscience and Nanotechnology 9(6): 3441-3448.
14. Bystrzycki J., Polański M., Malka I. E., Komuda A., 2009, *Hydriding properties of Mg - Al - Zn quasicrystal powder produced by mechanical alloying*, Zeitschriften fur Kristallographie 2241-2: 105-108.
15. Bystrzycki J., Płocioski T., Zielioski W., Wiśniewski Z., Polański M., Mróz W., Bojar Z., Kurzdłowski K. J., 2009, *Nano-engineering of magnesium hydride for hydrogen storage*, Microelectronic Engineering 86: 4-6: 889-891.
16. Polański M., Bystrzycki J., Plocinski T., 2008, *The effect of milling conditions on microstructure and hydrogen absorption/desorption properties of magnesium hydride (MgH₂) without and with Cr₂O₃ nanoparticles*, International Journal of Hydrogen Energy 33 (7): 1859-1867.
17. Bystrzycki J., Polański M., Perzyna K., 2007, *Badania materiałów do przechowywania wodoru w Wojskowej Akademii Technicznej*, Chemik 1: 19-20.